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for examiner use question mark - revision science - g/k93725/jun13/ch3hp 3 in this question you will be assessed on using good english, organising information clearly and using specialist terms where appropriate. ... g/k93725/jun13/ch3hp (13) 5 (b) describe how the student would test a solution of the low sodium salt for chloride ions.

gcse chemistry question paper unit 03 - chemistry june 2017 - g/jun17/ch3hp 2 (c) figure 3 shows apparatus used to measure the energy released when an alcohol is burned. figure 3 water glass beaker thermometer alcohol burner 2 (c) (i) in this question you will be assessed on using good english, organising information clearly and using specialist terms where appropriate.

ch3hp report on the examination - report on the examination gcse chemistry ch3hp june 2013 6 of 7 (c) (ii) this question was better answered although there was confusion with the limestone topic, resulting in answers relating to the production of carbon dioxide, scale or scum.

gcse chemistry question paper unit 03 - chemistry june 2014 - question 2(b)(ii) should be answered in continuous prose. in this question you will be marked on your ability to: use good english organise information clearly ... g/jun14/ch3hp (13) use the information in figure 7 and your own knowledge to answer this question.

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ch3fp report on the examination - question 7 on group 1 revealed that although students can be familiar with experiments they can have difficulty in understanding the chemistry involved. some

parts of question 8 on organic chemistry showed a lack of knowledge and understanding. question 9, requiring an extended answer on titration, was generally well answered.

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additional science ch2hp unit chemistry c2 h - additional science ch2hp unit chemistry c2 ... 13 do not write outside the box g/jun15/ch2hp (13) 5 (c) lead nitrate solution is produced by reacting lead oxide with nitric acid. ... 6 this question is about sodium chloride and iodine. 6 (a) describe the structure and bonding in sodium chloride.

mark scheme (results) summer 2013 international gcse ... - question number answer accept reject marks 3 (a) highest temperature temperature rise 28 3 30 6 32 9 32 9 1 mark for each column correct. mark temp. rise csq on highest temp.

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